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Solution Preview. 1) Which of the following statements is correct for an electron that has the quantum numbers $n = 4$ and $m_l = -2$? A. the electron may be in a p orbital B. the electron may be in a d orbital C. the electron is in the second principal shell D. the electron must have a spin quantum number $m_s = +\frac{1}{2}$

25 General Chemistry Multiple Choice Questions

Chemistry Principles of Modern Chemistry The vapor pressure of liquid mercury at 27°C is 2.87×10^{-6} atm . Calculate the number of Hg atoms per cubiccentimeter in the “empty” space above the top of the column of mercury in a barometer at 27°C .

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